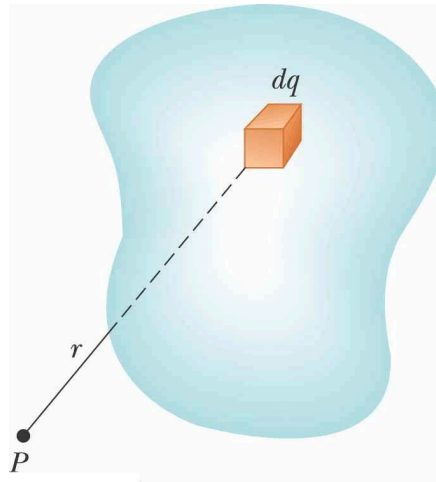


Continuous Charged Objects

- If the electric field is not known (or you are not sure how to find the electric field) for a continuous charged object, then we apply the superposition principle for all the charges making up the object in question.
- Breaking up the charged object into small portions of charge dq :

$$dV = \frac{k dq}{r}$$

$$V = \int_{\text{Body}} dV = \int \frac{k dq}{r}$$



PHYS102 - ★ ◀ ▶ □ ×

Electric Potential - Energy – slide 1

Reminder

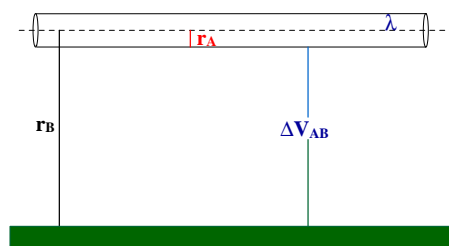
- Differences in potential energy - and thus in electric potential - have physical significance.
 - ◆ A reference potential is needed!
- Let's work out an example with a few different objects.

PHYS102 - ★ ◀ ▶ □ ×

Electric Potential - Energy – slide 2

Example - Power Line

- A long, straight power line is made from wire with radius $r_A = 1.0$ cm and carries a line charge density $\lambda = 2.6 \mu\text{C}/\text{m}$ as shown in the figure on the right. Assuming no other charges are present, what is the potential difference between the surface of the wire and the ground, a distance $r_B = 22$ m below?
 - i. Treat the wire as a very long wire, and apply Gauss's law to find the electric field:



$$\mathbf{E} = \frac{\lambda}{2\pi \epsilon_0 r} \hat{\mathbf{r}} \quad (r > r_A)$$

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Electric Potential - Energy – slide 3

Example - Power Line II

ii. Use the definition of potential difference.

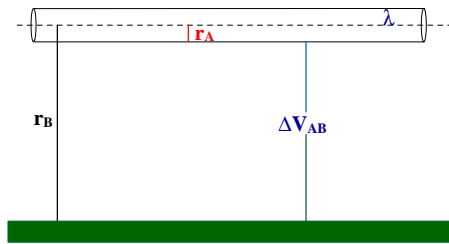
$$\Delta V_{AB} = - \int_{r_A}^{r_B} \vec{E} \cdot d\vec{l}$$

$$\Delta V_{AB} = - \int_{r_A}^{r_B} \frac{\lambda}{2\pi\epsilon_0 r} \hat{r} \cdot \hat{r} dr$$

$$\Delta V_{AB} = - \frac{\lambda}{2\pi\epsilon_0} \int_{r_A}^{r_B} \frac{dr}{r}$$

$$= \frac{\lambda}{2\pi\epsilon_0} \ln\left(\frac{r_A}{r_B}\right)$$

$$\Delta V = -360\text{kV}$$



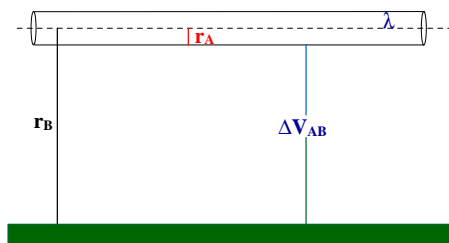
PHYS102 - ★ ◀ ▶ □ ×

Electric Potential - Energy – slide 4

Example - Power Line III

■ The potential difference is negative. WHY?

- ◆ Moving a positive charge q from r_A to r_B , would require a force opposite to the displacement.
- ◆ Moving a negative charge $-q$ from r_A to r_B , would require a force in the direction of the displacement.
- ◆ This is all summed up with the equation:

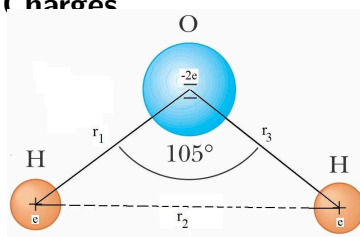


$$W = q\Delta V \quad \text{with } \Delta V = -360 \text{ kV}$$

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Electric Potential - Energy – slide 5

Example - Assembling Charges



■ Calculate the amount of work (energy) needed to assemble the three charges above (which is a crude approximation to the water molecule).

■ Let's assemble the bottom left charge ($+e$) first.

- ◆ Since there is no electric field initially, $W_{eL} = 0$.
- ◆ This charge sets up an electric potential, $V_1 = \frac{ke}{r}$

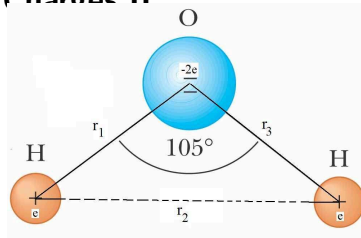
■ Assemble the $-2e$ charge next. This charge interacts with the potential generated by e , V_1 .

$$W_{2e} = -2eV_1(r_1) = -2e \frac{ke}{r_1} = -\frac{2ke^2}{r_1}$$

PHYS102 - ★ ◀ ▶ □ ×

Electric Potential - Energy – slide 6

Example - Assembling Charges II



- Assemble the last charge, $+e$ on the right hand side. This charge interacts with the two potentials, V_1 and V_2 , set up by the two other charges.

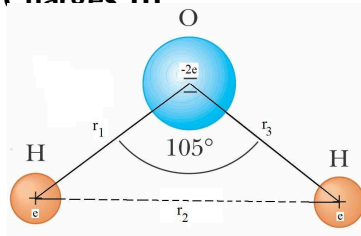
◆ $V_1 = \frac{k e}{r}$ and $V_2 = -\frac{2k e}{r}$.

- $W_{eR} = e(V_1(r_2) + V_2(r_3)) = \frac{k e^2}{r_2} - \frac{2k e^2}{r_3}$

- The total work required to assemble the charges in the configuration above is the sum of the above energies:

PHYS102 - ★ ◀ ▶ □ × $W_{\text{Total}} = W_{eL} + W_{2e} + W_{eR} = 0 - \frac{2k e^2}{r_1} + \frac{k e^2}{r_2} - \frac{2k e^2}{r_3}$ Electric Potential - Energy – slide 7

Example - Assembling Charges III



- Sum all the energies to find the total work required to assemble the charges.

$$W_{\text{Total}} = W_{eL} + W_{2e} + W_{eR} = 0 - \frac{2k e^2}{r_1} + \frac{k e^2}{r_2} - \frac{2k e^2}{r_3}$$

$$W_{\text{Total}} = k e^2 \left(-\frac{2}{r_1} + \frac{1}{r_2} - \frac{2}{r_3} \right) \text{ with } r_1 = r_3 = 0.1 \text{ nm}$$

PHYS102 - ★ ◀ ▶ □ × $W_{\text{Total}} = -7.76 \times 10^{-18} \text{ J} = -48.5 \text{ eV}$ Electric Potential - Energy – slide 8

Units

- On the last slide, we calculated a very small amount of energy in Joules. A more convenient unit of energy when dealing with atoms or molecules is the electron-Volt (eV) which is the amount of energy gained by a charge (e) passing through a potential difference of 1 V:

$$1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$$

PHYS102 - ★ ◀ ▶ □ × Electric Potential - Energy – slide 9

Potential Difference and \vec{E} -field

- Since the electric field and the potential difference are related by:

$$dV = -\vec{E} \cdot d\vec{l} = -E_l dl$$

$$E_l = -\frac{dV}{dl} \quad (\text{where } E_l \text{ is the component of the electric field parallel to } l)$$